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acid is building block
in manufacturing of;
Ethane C₂H₆, belongs to;
Non-polar side chains in
amino acids include;
Two optical isomers
are formed from
carbon; If cyclic-side
chain is bonded to

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both; Almond essence
in cakes and pudding
is made from

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Structures and

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structure. a) FCC b) BCC c) HCP d) TCP

View Answer

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9 multiple choice
questions. Each
question has four

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choices (a), (b), (c) and (d), out of which ONLY ONE is correct.

. (a) A Schottky defect consists of a pair of holes in the crystal lattice. One positive ion and one negative ion are absent.

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Two axes are the

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same, one different.
They are all different.
They are all the same.
Anything is possible
except for all three
the same. 3. In a
crystal structure only
4 types of symmetry
can be found. these
are? 21 fold, 32 fold,
12 fold, 42 fold. 1
fold, 5 fold, 7 fold, 9
fold. 2 fold, 3 fold, 4
fold, 6 fold.

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The edge of unit cell
of fcc Xe crystal is
620 pm.

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Sharpe. On opening the book cover you will find a periodic table and a list of elements and atomic masses.

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Questions

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The crystal structure
of iron is (A)
Simple Cubic (B) Face
centered cubic ...
hexagonal close
packed lattice (B)
body central cubic
lattice (C) simple
cubic lattice (D) none

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of the above. ANS : A.

10. Which of the following metal has lowest melting point?

... METALLURGY

Multiple Choice

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~~METALLURGY~~

~~Multiple Choice~~

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Questions and Answers

1. How many unit cells are divided equally in a face-centred cubic lattice?

2; 4; 6; 8; Answer: (c)

2. A crystalline solid: changes abruptly from solid to liquid when heated. has no definite melting point. undergoes deformation of its

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geometry easily; has
irregular
3-dimensional
arrangements.

Answer: (a) 3.

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7 18 Crystal
structure, Bravais
lattices, Crystal
system, unit cell,
Miller indices,
reciprocal lattice 3-

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Students will be able to determine the crystal structure by analysis of XRD data

5 8 X-ray, neutron, electron diffraction. Bragg's law in direct and reciprocal lattice. Structure factor. Diffraction techniques. 4- Students will be able to

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Deeper questions. The

following questions

require some thought

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and reaching the answer may require you to think beyond the contents of this TLP. Construct a plan view of NaCl (sodium chloride). NaCl has a face-centred cubic lattice. The motif is: Cl @ (0,0,0); Na @ (0,0,1/2); Note 1: The motif coordinates are positions relative to each lattice ...

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